

An Analysis on the Growth Effects of *Pseudomonas fluorescens* on Hexadecane Under Various Aquatic Conditions

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Introduction

As stated in Meghan Shea's paper, The Effect of Nitrogen, Sulfur, and Phosphorus Compounds on Bioremediation of Oil Spills by *Pseudomonas fluorescens* and *Bacillus subtilis*, marine oil spills are devastating and being able to prepare for them would limit damage. Our experiment was designed using Shea's fundamental idea for *Pseudomonas fluorescens* as an option for bioremediation, a technique in which organisms are used to remove pollutants from a place that is contaminated (1). The goal was to find which type of water created the best environment for the growth of *Pseudomonas fluorescens* (*P. fluorescens*) on hexadecane. Hexadecane, an aliphatic hydrocarbon, is a model representative of diesel fuel and our pollutant. By studying the growth of *P. fluorescens* on hexadecane, we were able to see how the organism's growth was affected based off the water used. It is probable that a water sample most like ocean water will provide the best environment for the growth of *P. fluorescens* on hexadecane.

Materials and Methods

Initially we needed to see how and if growth of *P. fluorescens* was inhibited by hexadecane. The 10x Luria Broth (LB) with tryptone and yeast extract provided enough nutrients where growth was not inhibited with the addition of hexadecane. A 250 μ l sample was made and 200 μ l was loaded into the Bioscreen C instrument, which was used to measure the optical density allowing for the construction of a growth curve. We depleted the nutrients, so *P. fluorescens* had to adapt or find other sources for nutrients. Once we lowered the nutrients to .5 μ l of both tryptone and yeast extract with 50 μ l of hexadecane varying water types was possible. Our water samples included filtered water from Lake Johnson in Raleigh, filtered instant ocean from Dr. James Brown's saltwater tank that will be referred to as Marine water, 1 M NaCl, tap water from the lab in Thomas Hall at NC State University, and distilled water used as a control.

Results

Based off maximum growth, Marine water provided the best conditions for the growth of *P. fluorescens* on hexadecane.

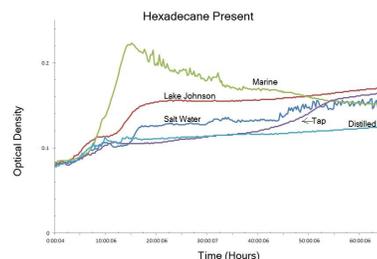


Fig. 1. Growth of *P. fluorescens* with hexadecane present of .5 μ l of tryptone and .5 μ l of yeast extract in various water to simulate different aquatic environments.

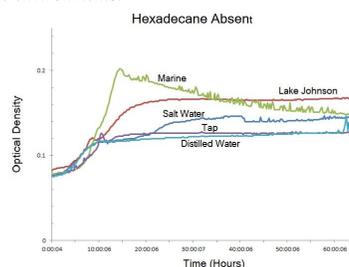


Fig. 2. Growth of *P. fluorescens* with no hexadecane present in .5 μ l of tryptone and .5 μ l of yeast extract in various water to simulate different aquatic environments.

Type of Water	Maximum Growth	
	Hexadecane Present	Hexadecane Absent
Salt Water (NaCl)	0.158	0.147
Lake Johnson	0.171	0.168
Marine Water	0.222	0.201
Tap Water	0.165	0.129
Distilled Water	0.125	0.149

Table 1. This shows the maximum growth of *P. fluorescens* in each of the different types of water.

Discussion

Although the water samples from Lake Johnson and the saltwater tank have unknown nutrients which allow us to have realistic data with our simulated aquatic environments, a large scale model of the reaction between *P. fluorescens* and hexadecane in different types of water would be more representative of a real world oil spill.

References

1) Shea, M. (2013). The effect of nitrogen, sulfur, and phosphorus compounds on bioremediation of oil spills by *pseudomonas fluorescens* and *bacillus subtilis*. The Journal of Experimental Secondary Science, 30-37.

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